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## Interim Report

IR-13-074

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1 **Nitrogen dynamics in Turbic Cryosols from Siberia and Greenland**

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31

## 32 **Abstract**

33 Turbic Cryosols (permafrost soils characterized by cryoturbation, i.e., by mixing of soil layers due to  
34 freezing and thawing) are widespread across the Arctic, and contain large amounts of poorly  
35 decomposed organic material buried in the subsoil. This cryoturbated organic matter exhibits  
36 retarded decomposition compared to organic material in the topsoil. Since soil organic matter (SOM)  
37 decomposition is known to be tightly linked to N availability, we investigated N transformation rates  
38 in different soil horizons of three tundra sites in north-eastern Siberia and Greenland. We measured  
39 gross rates of protein depolymerization, N mineralization (ammonification) and nitrification, as well  
40 as microbial uptake of amino acids and  $\text{NH}_4^+$  using an array of  $^{15}\text{N}$  pool dilution approaches. We  
41 found that all sites and horizons were characterized by low N availability, as indicated by low N  
42 mineralization compared to protein depolymerization rates (with gross N mineralization accounting  
43 on average for 14% of gross protein depolymerization). The proportion of organic N mineralized was

44 significantly higher at the Greenland than at the Siberian sites, suggesting differences in N limitation.  
45 The proportion of organic N mineralized, however, did not differ significantly between soil horizons,  
46 pointing to a similar N demand of the microbial community of each horizon. In contrast, absolute N  
47 transformation rates were significantly lower in cryoturbated than in organic horizons, with  
48 cryoturbated horizons reaching not more than 32% of the transformation rates in organic horizons.  
49 Our results thus indicate a deceleration of the entire N cycle in cryoturbated soil horizons, especially  
50 strongly reduced rates of protein depolymerization (16% of organic horizons) which is considered the  
51 rate-limiting step in soil N cycling.

52

53 Keywords: Arctic, tundra, cryoturbation, soil organic matter, ecological stoichiometry, nitrogen  
54 transformation, protein depolymerization, nitrogen mineralization, nitrification, nitrogen availability

55

56 **1. Introduction**

57 Arctic soils are commonly affected by cryoturbation, i.e., by a mixing of soil layers due to freeze-thaw  
58 processes, and are thus often characterized by horizons of poorly decomposed material subducted  
59 into and surrounded by mineral subsoil (Bockheim, 2007; Tarnocai et al., 2009). Cryoturbated soils  
60 are estimated to store 581 Gt of organic carbon that is currently protected from fast decomposition  
61 (Tarnocai et al., 2009). Although cryoturbated horizons ( $O_{jj}$  or  $A_{jj}$ ) are chemically similar to organic (O)  
62 or mineral topsoil (A) horizons, they are usually several hundred to thousand years older (Xu et al.,  
63 2009; Hugelius et al., 2010), suggesting that C mineralization is slowed down in cryoturbated  
64 horizons. Likewise, N mineralization has been found to be lower in cryoturbated than in all other  
65 horizons, i.e., in organic (O), mineral topsoil (A), and mineral subsoil (B) horizons (Kaiser et al., 2007).  
66 Cryoturbation thus lead to retardation of soil organic matter (SOM) decomposition in general, and of  
67 N mineralization in particular, and this retardation cannot be explained by lower subsoil  
68 temperatures alone (Kaiser et al., 2007).

69 The rate limiting step for soil N cycling is the breakdown of N-rich, high molecular weight organic  
70 compounds by extracellular enzymes, especially the depolymerization of proteins (Schimel and  
71 Bennett, 2004; Geisseler et al., 2010; Jones and Kielland, 2012). The resulting oligopeptides and  
72 amino acids are taken up by microorganisms and are further mineralized to  $NH_4^+$  which, in turn, is  
73 the substrate for nitrification (Jones and Kielland, 2012). Nitrogen mineralization is considered an  
74 overflow mechanism, i.e., only an excess of N that cannot be used to build up biomass (because  
75 another element is limiting) will be mineralized (Schimel and Bennett, 2004). Thus, gross N  
76 mineralization rates are expected to increase with increasing N availability, and, consequently,  
77 microbial N use efficiency (NUE; the proportion of N taken up that is incorporated into microbial  
78 biomass) will decrease.

79 Nitrogen availability in high latitude ecosystems is generally low (Schimel and Bennett, 2004), but is  
80 predicted to increase with global warming. Higher soil temperatures were shown to increase soil  
81 microbial activity, resulting in increased net N mineralization, and thus N availability (Nadelhoffer et  
82 al., 1991; Hobbie, 1996; Rustad et al., 2001). Additionally, the functional composition of tundra  
83 vegetation is expected to shift (Elmendorf et al., 2012), which is likely to alter the availability of N for  
84 soil microorganisms by changing patterns in plant N uptake, litter decomposability and competition  
85 for N in the rhizosphere (Wookey et al., 2009). An increase in N availability can stimulate plant and  
86 microbial growth (e.g., Hobbie et al., 2002; Sistla et al., 2012), and also SOM decomposition. In a  
87 long-term fertilization study in Alaskan tundra, Mack et al. (2004) observed that fertilization not only  
88 increased plant production, but also SOM decomposition, overall resulting in a net loss of C from the  
89 system. This suggests that the emission of CO<sub>2</sub> from Arctic soils could be amplified by increased N  
90 availability. The effect of N additions on SOM decomposition is, however, highly variable. Across  
91 sites, and across soil horizons, both stimulation and inhibition of SOM decomposition by N addition  
92 have been observed (Lavoie et al., 2011). To predict the overall effects of N availability on SOM  
93 decomposition in Arctic soils in a future climate, an in-depth understanding of all steps of N cycling in  
94 the soil profile is necessary. So far most studies have exclusively focused on N mineralization and  
95 nitrification, but the depolymerization step of high-molecular weight organic N, which has been  
96 suggested to be rate-limiting, has not received much attention.

97 While many studies have investigated N dynamics in organic and mineral topsoil horizons, not much  
98 is known about mineral and cryoturbated horizons in the subsoil. In spite of the large amounts of  
99 poorly decomposed SOM in cryoturbated horizons, and in spite of the importance of N for SOM  
100 decomposition, we are only aware of one study investigating N dynamics in cryoturbated horizons.  
101 This study showed a reduction in N mineralization rates in cryoturbated compared to organic  
102 horizons (Kaiser et al., 2007). These low N mineralization rates could either have been the  
103 consequence of reduced protein depolymerization rates, or of a higher N demand of the microbial

104 community. In the latter case, protein depolymerization would be similar in all horizons, but N  
105 mineralization would be lower in cryoturbated horizons, indicating that the microbial community  
106 needed a higher proportion of the available N for growth, and thus had a higher NUE.

107 We here report on microbial N transformations and NUE across different horizons in the active layer  
108 of tundra soil. We hypothesized that the observed reduction of gross N mineralization in  
109 cryoturbated horizons compared to organic horizons was due to a reduction in protein  
110 depolymerization, not due to a higher NUE of the microbial community. Since the capacity to  
111 depolymerize proteins constitutes a property of the microbial community, we further investigated  
112 the microbial community structure in different soil horizons, with the goal to identify groups that  
113 may be responsible for individual N transformations.

114 To achieve these goals, we sampled Turbic Cryosols (Turbels) at three sites in Siberia and Greenland  
115 and measured gross rates of protein depolymerization, microbial amino acid uptake, N  
116 mineralization,  $\text{NH}_4^+$  uptake and nitrification using an array of  $^{15}\text{N}$  pool dilution approaches.  
117 Additionally, we estimated the microbial community composition using phospholipid fatty acids  
118 (PLFAs) as biomarkers.

119

## 120 **2. Material & Methods**

### 121 **2.1. Sampling Sites**

122 We compared soils from three different sites: (1) The heath tundra site was located in eastern  
123 Greenland close to the Zackenberg Research Station (typical tundra subzone; 74°29' N, 20°32' W) on  
124 sedimentary bedrock (sandstone). It was dominated by *Cassiope tetragona*, *Vaccinium uliginosum*,  
125 *Dryas octopetala*, *Salix arctica*, and *Carex* sp., with lichens between the dwarf shrubs. (2) The shrub  
126 tundra site was in north-eastern Siberia close to the town of Chersky (southern tundra subzone;

127 68°45' N, 161°36' E) on aeolian late Pleistocene sediment. It was located in a shrubby moss lichen  
128 tundra dominated by *Betula exilis*, *Vaccinium uliginosum*, *Flavocetraria nivalis*, *F. cucullata* and  
129 *Aulacomnium turgidum*, with sparse *Larix gmelinii* trees. (3) The tussock tundra site was  
130 approximately 80 km north of Chersky (southern tundra subzone; 69°26' N, 161°44' E) on aeolian late  
131 Pleistocene sediment. It was dominated by *Eriophorum vaginatum*, *Carex lugens*, *Betula exilis*,  
132 *Vaccinium vitis-idaea*, *Aulacomnium turgidum*, and *Dicranum* sp.

133 All soils were classified as Turbic Cryosols according to the World Reference Base for Soil Resources  
134 (IUSS Working Group WRB, 2007) or Turbels according to the US Soil Taxonomy (Soil Survey Staff,  
135 1999). The active layer at the time of sampling was 47 cm (heath tundra, Greenland), 73 cm (shrub  
136 tundra, Siberia; under frost boils) and 72 cm (tussock tundra, Siberia; under frost boils).

137 At each site, we took samples from the active layer of three replicate soil pits. We sampled organic  
138 layers (O, including OA horizons) from the soil surface (further termed "organic horizon"). We then  
139 took samples from pockets of organic material (O<sub>ij</sub>) or mineral topsoil (A<sub>ij</sub>) that was buried within the  
140 subsoil. Such buried horizons, caused by cryoturbations, i.e., by subduction of organic material or  
141 mineral topsoil into the subsoil due to freeze-thaw processes (Bockheim, 2007), are common in  
142 permafrost soils. Sampled pockets of cryoturbated material were between 20 and 50 cm from the  
143 soil surface (further termed "cryoturbated horizon"). We further sampled the mineral soil  
144 surrounding the cryoturbated pockets (A, AB, B or C<sub>g</sub> horizons, at depths from 10 to 60 cm from the  
145 surface). These horizons are collectively termed "mineral horizon". Living roots were carefully  
146 removed and samples were kept cool until analysis.

147 All samplings took place in August 2010. We are aware of the fact that N cycling and N availability  
148 show seasonal variation in Arctic soils (e.g., Weintraub and Schimel, 2005a; Weintraub and Schimel,  
149 2005b; Edwards et al., 2006). Data presented here do not reflect these variations, but represent N  
150 cycling in the late growing season.

151

## 152 **2.2. Basic Characterization and Nutrient Concentrations**

153 Concentrations of ammonium and nitrate were determined photometrically in 1 M KCl extracts  
154 following Kandeler and Gerber (1988) and Miranda et al. (2001), respectively. Total dissolved  
155 nitrogen (TDN) was measured with a DOC/TN analyzer (Elementar LiquiToC II or Shimadzu TOC-  
156 V<sub>CPH/CPN/TNM-1</sub>) in 0.5 M K<sub>2</sub>SO<sub>4</sub> or 1 M KCl extracts (samples from Siberia and Greenland,  
157 respectively), and dissolved organic N (DON) was calculated by subtracting NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> from TDN.  
158 Extraction with K<sub>2</sub>SO<sub>4</sub> or KCl was previously shown to yield similar DON recovery (Jones and Willett,  
159 2006). For determination of inorganic phosphate, soil samples were extracted with 0.5 M NaHCO<sub>3</sub>  
160 (Olsen et al., 1954) and measured photometrically with the molybdate-ascorbic acid method  
161 (Murphy and Riley, 1962). Contents of organic C and total N were determined in dried and ground  
162 samples using elemental analysis-isotope ratio mass spectrometry (EA-IRMS), either with a CE  
163 Instrument EA 1110 elemental analyzer coupled to a Finnigan MAT DeltaPlus IRMS with a Finnigan  
164 MAT ConFlo II Interface, or with an Isoprime EA-IRMS system. Samples from Siberia contained traces  
165 of carbonate and were acidified in HCl atmosphere and neutralized over NaOH before EA-IRMS  
166 analysis. For determination of total P content, samples were amended with a mixture of  
167 concentrated HClO<sub>4</sub> and HNO<sub>3</sub> (1:4), heated stepwise to 160°C and 220°C for digestion, cooled to  
168 room temperature, filtered (Whatman 40 ashfree cellulose filter) and measured with inductively  
169 coupled plasma-optical emission spectrometry (ICP-OES, Perkin Elmer Optima 3000 XL) against  
170 external standards. Ratios of C, N and P were calculated as mass ratios. pH values were determined  
171 in suspensions of dried soil in de-ionized water (1:2.5; weight:volume).

172

## 173 **2.3. Gross Rates of N Transformations**

174 Gross rates of protein depolymerization and amino acid uptake were determined using a  $^{15}\text{N}$  pool  
175 dilution assay as described by Wanek et al. (2010), with slight modifications to account for the low  
176 amino acid concentrations in soils. Briefly, a mixture of 20  $^{15}\text{N}$  labeled amino acids (>98 at%, Spectra  
177 and Cambridge Isotope Laboratories) was dissolved in 10 mM  $\text{CaSO}_4$  and added to 2 g of field-moist  
178 soil in duplicates. Per sample, 500  $\mu\text{l}$  solution containing 2.5  $\mu\text{g}$  total amino acids were applied. After  
179 incubation for 10 or 30 min at 7°C, activities were stopped with 19.5 ml of 10 mM  $\text{CaSO}_4$  containing  
180 3.7% of formaldehyde. Samples were extracted for 5 min and either allowed to settle for 10 min  
181 (both Siberian sites) or centrifuged for 5 min at 10 845 g (Greenland heath tundra site). Samples were  
182 filtered through synthetic wool and GF/C filters (Whatman), and loaded on cation exchange  
183 cartridges (OnGuard II H 1cc cartridges, Dionex, cleaned with 3M ammonia and re-generated with 1  
184 M HCl before use). After application of the samples, cartridges were washed with 10 ml distilled  
185 water, stabilized with 5 ml 5% methanol and stored cool until elution. With each batch of samples,  
186 blanks and amino acid standards were processed to correct for losses due to ion exchange. After  
187 elution of the amino acids from the cartridges with 30 ml 3M ammonia, an internal standard was  
188 added to the samples (1  $\mu\text{g}$  of nor-valine, nor-leucine and para-chloro-phenylalanine each, Sigma  
189 Aldrich). Samples were dried with rotary evaporation, re-dissolved in 1.5 ml 20% ethanol and dried in  
190 a SpeedVac system. Finally, samples were derivatized with chloroformate before analysis with a gas  
191 chromatography-mass spectrometry system (GC-MS), consisting of a CTC autosampler (CTC Analytics)  
192 and a Trace GC Ultra coupled to a quadrupole mass spectrometer (DSQ II; Thermo Scientific). Two  $\mu\text{l}$   
193 of sample were injected in splitless mode (injector temperature 270°C), separated on an Equity-1701  
194 column (30 m x 0.25 mm x 1  $\mu\text{m}$ ; Sigma-Aldrich) with 1 ml  $\text{min}^{-1}$  He as carrier gas (GC method: 105°C  
195 for 1 min, 6°C  $\text{min}^{-1}$  to 135°C, 3°C  $\text{min}^{-1}$  to 180°C, 20°C  $\text{min}^{-1}$  to 260°C, 260°C for 35 min) and detected  
196 in Selected Ion Monitoring mode. Concentrations of alanine, valine, leucine, isoleucine, proline,  
197 tryptophane, phenylalanine and tyrosine were calculated using calibrations against external  
198 standards, and the abundance of  $^{15}\text{N}$  in each of these amino acids was calculated based on peak

199 areas of fragments characteristic for  $^{14}\text{N}$  and  $^{15}\text{N}$  (for fragments see Wanek et al., 2010), using a  
200 calibration against standards of different  $^{15}\text{N}$  abundance.

201 Gross rates of N mineralization (ammonification),  $\text{NH}_4^+$  uptake and nitrification were determined as  
202 described by Kaiser et al. (2011) by adding 500  $\mu\text{l}$   $^{15}\text{N}$  labeled  $(\text{NH}_4)_2\text{SO}_4$  (N mineralization and  $\text{NH}_4^+$   
203 uptake) or  $\text{KNO}_3$  (nitrification) to duplicates of 2 g of field-moist soil (0.125 mM, 10 at%, Sigma  
204 Aldrich). Samples were incubated for 4 h and 24 h at  $7^\circ\text{C}$ , extracted with 13 ml of 2 M KCl for 30 min,  
205 and filtered through ashfree filter paper (Whatman 40 ashfree cellulose filter). To stabilize the  
206 extracts, 20  $\mu\text{l}$  5 mM phenylmercuric acetate were added, and samples were frozen until further  
207 processing. For N mineralization and  $\text{NH}_4^+$  uptake,  $\text{NH}_4^+$  was diffused into acid traps and measured  
208 with an EA-IRMS system consisting of a CE Instrument EA 1110 elemental analyzer coupled to a  
209 Finnigan MAT DeltaPlus IRMS with a Finnigan MAT ConFlo II Interface. For nitrification,  $\text{NH}_4^+$  was  
210 removed from the samples and  $\text{NO}_3^-$  converted to  $\text{NH}_4^+$  before diffusion into acid traps and EA-IRMS  
211 analysis (Mooshammer et al., 2012). Gross rates were based on differences in concentration and  
212 isotopic composition of  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  or amino acids between two time points (e.g., 4 h and 24 h) and  
213 were calculated according to the equations described in Wanek et al. (2010).

214 As an indicator for microbial N limitation, we calculated the efficiency of microorganisms to use  
215 amino acid N for biomass growth (N use efficiency, NUE) by comparing gross rates of amino acid  
216 uptake and N mineralization:

$$217 \text{ NUE} = (\text{gross amino acid uptake} - \text{gross N mineralization}) / \text{gross amino acid uptake}.$$

218

#### 219 **2.4. Phospholipid Fatty Acid (PLFA) Analysis**

220 For analysis of PLFAs, samples were stored frozen (Greenland heath tundra) or in RNAlater (both  
221 Siberian sites; Schnecker et al., 2012). Phospholipid Fatty Acids were extracted from 1 g of soil with

222 chloroform/methanol/citric acid buffer and purified on silica columns (LC-Si SPE, Supelco) using  
223 chloroform, acetone and methanol (Frostegård et al., 1991; with the modifications described by  
224 Kaiser et al., 2010). After addition of methyl-nonadecanoate as internal standard, PLFAs were  
225 converted to fatty acid methyl esters (FAMES) by alkaline methanolysis. Samples were analyzed on a  
226 Thermo Trace GC with FID detection: 1 µl per sample was injected in splitless mode (injector  
227 temperature 230°C) and separated on a DB-23 column (Agilent; GC method:70°C for 1.5 min, 30°C  
228 min<sup>-1</sup> to 150°C, 150°C for 1 min, 4°C min<sup>-1</sup> to 230°C, 230°C for 15 min) with 1.5 ml min<sup>-1</sup> He as carrier.  
229 Individual FAMES were identified using qualitative standard mixes (37 Comp. FAME Mix and Bacterial  
230 Acid Methyl Esters CP Mix, Supelco) and quantified by comparison with the internal standard. We  
231 used 18:1ω9, 18:2ω6,9 and 18:3ω3,6,9 fatty acids as biomarkers for fungi, i15:0, a15:0, i16:0, i17:0  
232 and a17:0 for gram positive bacteria, cy17:0 (9/10), cy19:0 (9/10), 16:1ω5, 16:1ω7, 16:1ω9 and  
233 18:1ω7 for gram negative bacteria, and 14:0, 15:0, 16:0, 17:0, 18:0, 20:0, i14:0, 16:1ω10, 16:1ω11,  
234 17:1ω6 and 10Me16:0 as unspecific markers (Kaiser et al., 2010).

235

## 236 **2.5. Statistics**

237 To test for significant differences between sites and horizons, we applied two-way ANOVAs and  
238 Tukey HSD tests (after transformation, if necessary) or Kruskal-Wallis tests with unpaired Mann  
239 Whitney U tests as post-hoc tests if normal distribution and homoscedasticity could not be achieved.  
240 We additionally performed a Principal Component Analysis including gross N transformation rates  
241 (per g total N), pH values and the relative abundance of microbial groups. Samples with missing  
242 values were omitted from the Principal Component Analysis. Correlations were tested using  
243 Spearman's rank correlations. All statistics were performed in R 2.15 (R Development Core Team,  
244 2012).

245

246 **3. Results**

247 Organic C, total N, and total P, as well as the C/N ratio, decreased significantly from organic to  
248 cryoturbated and mineral horizons (Tables 1 and 2). pH-Values were in the range of 4.3 to 5.5 for  
249 organic horizons and significantly higher in cryoturbated and mineral horizons (5.1-6.1 and 5.3-6.4,  
250 respectively). Microbial biomass (estimated as the total amount of PLFAs per g DW) decreased  
251 significantly from organic to cryoturbated and mineral horizons (Supplementary Fig. 1, Table 2),  
252 mainly due to differences in SOM content.

253 In all horizons, the pool of dissolved N was dominated by organic N (Fig. 1). The relative contribution  
254 of DON to the TDN pool was similar in all horizons at each site, although absolute concentrations  
255 were significantly lower in mineral than in organic horizons (Table 2). We did, however, observe  
256 significant variations in the composition of the TDN pool between sites. Dissolved organic N  
257 accounted for  $85 \pm 3\%$  (mean  $\pm$  standard error of all horizons) and  $86 \pm 5\%$  of the TDN pool at the  
258 heath tundra and tussock tundra sites, but only for  $58 \pm 7\%$  at the shrub tundra site (Fig. 1). Also  
259 concentrations of inorganic phosphate varied significantly between sites. Inorganic phosphate was  
260 lower by a factor of ten at the heath tundra site in Greenland than at both Siberian sites, but did not  
261 show significant differences between horizons (Fig. 1).

262 Gross rates of protein depolymerization, amino acid uptake, N mineralization,  $\text{NH}_4^+$  uptake and  
263 nitrification, expressed per g DW, were generally highest in organic horizons, and decreased  
264 significantly from organic to cryoturbated and mineral horizons (Supplementary Fig. 2, Table 2). All N  
265 transformation rates were significantly and positively correlated with C and N content (Table 3). In  
266 order to assess horizon-specific differences in transformation rates that were not related to  
267 differences in SOM content, we calculated all gross N transformation rates per g total soil N.  
268 Cryoturbated horizons still exhibited significantly lower gross rates of protein depolymerization,  
269 amino acid uptake, N mineralization and nitrification per g total N than organic horizons, on average

270 accounting for 16% (protein depolymerization), 32% (amino acid uptake), 27% (N mineralization) and  
271 31% (nitrification) of the respective rates in organic horizons. In the case of N mineralization and  
272 nitrification, rates in cryoturbated horizons were even significantly lower than in mineral horizons  
273 (Fig. 2, Table 2).

274 In spite of the differences in absolute N transformation rates, NUE did not differ significantly  
275 between horizons (Table 2), and was not correlated with soil C/N ratio ( $p=0.130$ ,  $R^2=0.10$ ). We did,  
276 however, observe significant differences in NUE between sites. NUE was highest at the tussock  
277 tundra site in Siberia, where 90% of amino acid N taken up was incorporated into microbial biomass,  
278 and only 10% were mineralized to  $\text{NH}_4^+$ . NUE was lower at both other sites, with 66% incorporation  
279 at the shrub tundra site in Siberia, and 51% at the heath tundra site in Greenland (Fig. 4).

280 To investigate possible relationships between N transformations and microbial community  
281 composition across horizons and sites, we performed a Principle Component Analysis, including gross  
282 N transformation rates (per g total N), the relative abundances of fungi, gram negative and gram  
283 positive bacteria (in % of total PLFAs), and pH values (Fig. 3). Principal Component 1 accounted for  
284 47% of the variation in the data set, and was positively connected to all gross N transformation rates  
285 (factor loadings 0.42 for protein depolymerization, 0.38 for amino acid uptake, 0.21 for N  
286 mineralization, 0.27 for  $\text{NH}_4^+$  uptake and 0.38 for nitrification), and thus represents an aggregated  
287 parameter of N transformation activity. Principal Component 1 separated all cryoturbated horizons  
288 and the mineral horizons of the Greenland heath tundra site from organic and the Siberian mineral  
289 horizons. These findings were supported by a two-way ANOVA that showed significantly lower values  
290 of Principal Component 1 for cryoturbated horizons compared to organic and mineral horizons, and  
291 for the Greenland heath tundra site compared to both Siberian sites (Table 2). Principal Component 2  
292 (16% of variation), in contrast, was positively connected to gross rates of protein depolymerization  
293 and microbial uptake of amino acids (factor loadings 0.13 and 0.19), but negatively connected to  
294 gross rates of N mineralization,  $\text{NH}_4^+$  uptake and nitrification (-0.55, -0.43 and -0.31), thus separating

295 organic from inorganic N transformation processes. Principal Component 2 was also significantly  
296 correlated with NUE which reflects the allocation of amino acid N to mineralization ( $p=0.004$ ,  
297  $R^2=0.34$ ).

298 The microbial groups of fungi, gram negative and gram positive bacteria contributed differently to  
299 Principal Components 1 and 2. Principal Component 1 was positively connected to the relative  
300 abundances of fungi and gram negative bacteria (factor loadings 0.37 and 0.27), and negatively to  
301 gram positive bacteria (-0.30). Principal Component 2 was positively connected to fungi (0.26) and  
302 negatively to both gram negative and gram positive bacteria (-0.27 and -0.35).

303 We further investigated possible relationships between individual N transformation processes and  
304 microbial groups using Spearman's rank correlations, and found that gross rates of protein  
305 depolymerization, amino acid uptake and nitrification (per g total N) were significantly correlated  
306 with the relative abundance of fungi, and that gross rates of amino acid uptake and nitrification were  
307 significantly correlated with gram negative bacteria (Table 3).

308

#### 309 **4. Discussion**

310 High-latitude systems are usually characterized by low N availability that limits both plant and  
311 microbial growth (Hobbie et al., 2002; Sistla et al., 2012). Plants and microorganisms rapidly  
312 immobilize all reactive N forms small enough for uptake, in particular amino acids (Jones and  
313 Kielland, 2002; Näsholm et al., 2009) and oligo-peptides (Hill et al., 2011; Farrell et al., 2013), and  
314 incorporate the N into their biomass, with a minimum of microbial overflow mineralization to  $\text{NH}_4^+$ ,  
315 and further transformation to  $\text{NO}_3^-$ . Although organic N forms represent a major source of N for both  
316 plants and microorganisms in the Arctic, we know little about the steps controlling their availability,  
317 i.e., protein depolymerization.

318

319 **4.1. Nitrogen cycling in cryoturbated horizons**

320 In the Arctic soils studied, gross rates of protein depolymerization, but also microbial amino acid  
321 uptake, N mineralization and nitrification were significantly lower in cryoturbated than in organic  
322 horizons, on average accounting for only 26% of the rates in organic horizons (Fig. 2). This  
323 corroborates our hypothesis that the whole sequence of N transformations, starting with the rate-  
324 limiting step of protein depolymerization, is decelerated in cryoturbated compared to organic  
325 horizons.

326 Protein depolymerization limits the amount of amino acids available for microbial uptake, and with it  
327 the potential for microbial growth and N mineralization. Severe N deficiency can even limit the  
328 production of extracellular enzymes that depolymerize complex organic compounds (Weintraub and  
329 Schimel, 2005a; Wallenstein et al., 2009; Sistla et al., 2012), including N-containing macromolecules  
330 such as proteins. While we did not directly estimate N availability in cryoturbated horizons, lower  
331 protein depolymerization rates compared to organic horizons point to a reduced N availability in  
332 cryoturbated horizons. We therefore suggest that the slow decomposition of cryoturbated SOM  
333 (Kaiser et al., 2007; Xu et al., 2009; Hugelius et al., 2010) might be connected to N limitation of  
334 enzyme production. In this case, an increase in protein depolymerization with climate change  
335 (Weedon et al., 2011; Brzostek et al., 2012) could facilitate the decomposition of cryoturbated SOM,  
336 and lead to higher CO<sub>2</sub> emissions from Arctic soils.

337

338 **4.2. Microbial communities and nitrogen transformations**

339 The differences in N cycling across soil horizons and sites were likely caused, at least in part, by  
340 differences in composition and N demand of the microbial communities. The rate-limiting step in N

341 cycling, protein depolymerization, requires specific enzymes, i.e., proteases, that are produced by a  
342 range of bacteria and fungi. We here found that gross protein depolymerization rates were  
343 significantly correlated with the relative abundance of fungi (Table 3). Fungi are able to produce a  
344 wide range of extracellular enzymes (Baldrian et al., 2011; Schneider et al., 2012) and are involved in  
345 the degradation of many complex organic molecules including cellulose and lignin (de Boer et al.,  
346 2005; Strickland and Rousk, 2010). Protein breakdown in particular has been assigned to ecto- and  
347 ericoid mycorrhizal fungi in high latitude systems (Read and Perez-Moreno, 2003). A low abundance  
348 of fungi in cryoturbated horizons might thus not only contribute to the low protein depolymerization  
349 rates, but also generally to the retarded decomposition of cryoturbated SOM.

350 Protein depolymerization was closely correlated with microbial amino acid uptake (Table 3). We  
351 found a similar close correlation for N mineralization and  $\text{NH}_4^+$  uptake. This suggests that the  
352 microbial uptake of amino acids and  $\text{NH}_4^+$  was limited by the respective production rates (protein  
353 depolymerization and N mineralization), indicating a high demand of the microbial biomass for N. A  
354 tight coupling of production and consumption rates of both amino acids and  $\text{NH}_4^+$  has already been  
355 demonstrated for decomposing beech leaf litter (Mooshammer et al., 2012).

356 While N mineralization obviously limited the amount of  $\text{NH}_4^+$  available for microbial uptake, N  
357 mineralization itself was not correlated with any upstream process such as protein depolymerization  
358 or amino acid uptake, or with any microbial group (Table 3). Nitrogen mineralization is the microbial  
359 de-amination of organic N and excretion of  $\text{NH}_4^+$ . The potential for N mineralization thus depends on  
360 the uptake of organic N by microorganisms. Microorganisms, however, can directly control the  
361 amount of N mineralized, and will only mineralize an excess of N that is not needed for growth or  
362 other cellular processes (Schimel and Bennett, 2004). Actual N mineralization rates therefore reflect  
363 both N availability and N demand for growth by the microbial community.

364 Nitrification, in contrast, was significantly correlated with the relative abundances of gram negative  
365 bacteria and fungi (Table 3). Nitrification requires a specific set of enzymes that oxidize ammonium  
366 over nitrite to nitrate, and is restricted to specific microbial groups. Autotrophic nitrification was  
367 found in few groups of archaea and gram negative bacteria (Hayatsu et al., 2008; Schleper, 2010;  
368 Alves et al., 2013). The abundance of gram negative bacteria might therefore influence nitrification  
369 rates, as already demonstrated for savanna and forest soils (Balsler and Firestone, 2005).  
370 Heterotrophic nitrification, in contrast, is more widespread among microorganisms, but was mainly  
371 connected to fungi, particularly in acidic soils (Hayatsu et al., 2008).

372

### 373 **4.3. Nitrogen use efficiency of the microbial community**

374 In addition to low protein depolymerization rates in cryoturbated horizons, the observed reduction in  
375 N mineralization may also have been caused by a higher allocation of the available N to growth, and  
376 thus by a higher NUE. Our results demonstrate, however, that microbial NUE did not differ  
377 significantly between horizons (Fig. 4), indicating that the microbial communities in all horizons,  
378 including the cryoturbated ones, had a similar demand for N. Furthermore, NUE was not correlated  
379 with soil C/N ratios, suggesting that the decrease in C/N ratio from organic to mineral horizons (Table  
380 1) was likely offset by changes in C and N availability with soil depth. Montané et al. (2007) showed  
381 for a mountain grassland soil, that while persistence of soil C was constant across the soil profile,  
382 persistence of soil N increased with soil depth, probably due to differences in chemical composition  
383 or in binding to soil minerals (for a recent review on SOM persistence see Schmidt et al., 2011).

384 NUE was generally rather high (as expected for N-limited systems), but differed significantly between  
385 sites (Fig. 4). NUE was significantly higher and gross N mineralization significantly lower at the  
386 tussock tundra site in Siberia than at the heath tundra site in Greenland, indicating that the microbial  
387 community at the tussock tundra site needed a higher proportion of the available N for growth. Size

388 and composition of the TDN pool, as well as gross protein depolymerization and amino acid uptake  
389 rates were similar at both sites (Table 2), so it can be assumed that N availability for microorganisms  
390 was similar – pointing to the fact that N was not the main limiting element at the Greenland heath  
391 tundra site.

392 Since concentrations of inorganic phosphate were significantly lower at the Greenland heath tundra  
393 than at both Siberian sites (Fig. 1), we suggest that microbial growth at the Greenland heath tundra  
394 site could have been rather limited in P. Microorganisms have to maintain a rather constrained C:N:P  
395 ratio and have only limited capacities to store excess C, N or P in the biomass. Therefore, P limitation  
396 should lead to increased N mineralization as an overflow mechanism (Tezuka, 1990; Sterner and  
397 Elser, 2002). Although P limitation is usually considered characteristic for old, highly weathered soils,  
398 for example in the tropics (Cleveland et al., 2002), there is increasing evidence that P is an important  
399 co-limiting and in some cases even the main limiting element in Arctic soils (Shaver and Chapin, 1995;  
400 Shaver et al., 1998; Giesler et al., 2002; Hartley et al., 2010; Giesler et al., 2012). In the case of the  
401 heath tundra site, we found that although concentrations of inorganic phosphate were significantly  
402 lower than at the Siberian sites, total P content was similar, indicating that differences in the sorption  
403 of P to mineral soil particles may have been responsible for the low P availability at this site.

404

#### 405 **4.4. Conclusions**

406 We found significant differences in N transformation rates and microbial N use efficiency between  
407 sites and horizons, demonstrating that the respective microbial communities differed in nutrient  
408 limitation. Across all sites, N cycling was slower in cryoturbated compared to organic horizons,  
409 starting with protein depolymerization, which is rate-limiting in N cycling. Our results thus indicate  
410 that microbial communities have a lower capacity to break down proteins in cryoturbated compared  
411 to organic horizons, likely due to differences in community composition (e.g., a lower abundance of

412 fungi). Overall, our study suggests that burial of organic material by cryoturbation into the subsoil  
413 leads to changes in soil N transformations, which in turn may contribute to the observed retarded  
414 decomposition of cryoturbated SOM by altering N availability for microbial decomposers.

415

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420

421 **Tables**

Table 1. C, N and P content, C/N and N/P ratios (mass ratios), and pH values of organic, cryoturbated (cryot.) and mineral horizons of heath tundra (Greenland), tussock tundra (Siberia) and shrub tundra (Siberia). Values represent means ( $\pm$  standard error).

site	horizon	C (%)	N (%)	P (%)	C/N	N/P	pH
heath	organic	15.76 (0.86)	0.93 (0.06)	0.102 (0.006)	17.14 (1.95)	9.18 (0.39)	5.29 (0.10)
	cryot.	9.35 (0.95)	0.65 (0.00)	0.112 (0.007)	14.46 (1.46)	5.82 (0.37)	6.02 (0.02)
	mineral	3.94 (1.01)	0.29 (0.08)	0.074 (0.025)	13.81 (0.61)	4.86 (1.57)	6.16 (0.11)
tussock	organic	22.15 (1.57)	1.00 (0.05)	0.155 (0.019)	22.38 (2.18)	6.68 (1.09)	5.13 (0.05)
	cryot.	15.30 (4.84)	0.81 (0.18)	0.112 (0.011)	18.23 (1.76)	7.36 (1.81)	5.57 (0.28)
	mineral	2.08 (0.40)	0.14 (0.02)	0.046 (0.039)	14.33 (0.40)	9.40 (7.42)	5.50 (0.02)
shrub	organic	26.63 (5.98)	0.96 (0.18)	0.115 (0.005)	27.46 (1.61)	8.36 (1.57)	4.64 (0.23)
	cryot.	9.69 (2.57)	0.64 (0.20)	0.121 (0.084)	15.46 (0.88)	8.01 (3.90)	5.80 (0.29)
	mineral	2.50 (0.85)	0.16 (0.03)	0.061 (0.022)	14.66 (2.96)	4.55 (2.66)	5.54 (0.19)

422

423

Table 2. Significance of differences between sites (heath tundra, Greenland; tussock tundra, Siberia; shrub tundra, Siberia) or between horizons (organic; cryoturbated; mineral), derived from two-way ANOVA with Tukey HSD test, or Kruskal-Wallis tests with Mann-Whitney U tests as post-hoc. Different letters indicate  $p < 0.05$ , with "a" denoting the highest values.

Levels of significance: \*\*\*,  $p < 0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ ; n.s., not significant.

parameter	unit	between sites			between horizons				
		sign.	heath	tussock	shrub	sign.	organic	cryot.	mineral
C	%	n.s.				***	a	b	c
N	%	n.s.				***	a	a	b
P	%	n.s.				*	a	ab	b
C/N	$g\ C\ g^{-1}\ N$	n.s.				***	a	b	b
N/P	$g\ N\ g^{-1}\ P$	n.s.				n.s.			
pH		**	a	b	b	***	b	a	a
Total PLFAs	$\mu\text{mol}\ g^{-1}\ DW$	n.s.				***	a	b	c
TDN	$\mu\text{g}\ N\ g^{-1}\ DW$	n.s.				***	a	b	b
DON	$\mu\text{g}\ N\ g^{-1}\ DW$	n.s.				**	a	b	b
$NH_4^+$	$\mu\text{g}\ N\ g^{-1}\ DW$	*	b	ab	a	*	a	ab	b
$NO_3^-$	$\mu\text{g}\ N\ g^{-1}\ DW$	n.s.				n.s.			
DON	% of TDN	**	a	a	b	n.s.			
Inorganic phosphate	$\mu\text{g}\ P\ g^{-1}\ DW$	**	b	a	a	n.s.			
Gross protein depolymerization	$\mu\text{g}\ N\ g^{-1}\ DW\ d^{-1}$	n.s.				***	a	b	b
Gross amino acid uptake	$\mu\text{g}\ N\ g^{-1}\ DW\ d^{-1}$	n.s.				***	a	b	b
Gross N mineralization	$\mu\text{g}\ N\ g^{-1}\ DW\ d^{-1}$	***	a	b	a	***	a	b	b
Gross $NH_4^+$ uptake	$\mu\text{g}\ N\ g^{-1}\ DW\ d^{-1}$	*	ab	b	a	***	a	a	b
Gross nitrification	$\mu\text{g}\ N\ g^{-1}\ DW\ d^{-1}$	n.s.				**	a	b	b
Gross protein depolymerization	$mg\ N\ g^{-1}\ N\ d^{-1}$	*	b	ab	a	*	a	b	ab
Gross amino acid uptake	$mg\ N\ g^{-1}\ N\ d^{-1}$	n.s.				*	a	b	ab
Gross N mineralization	$mg\ N\ g^{-1}\ N\ d^{-1}$	***	a	b	a	***	a	b	a
Gross $NH_4^+$ uptake	$mg\ N\ g^{-1}\ N\ d^{-1}$	**	b	b	a	n.s.			
Gross nitrification	$mg\ N\ g^{-1}\ N\ d^{-1}$	**	b	ab	a	*	a	b	a
NUE		**	b	a	b	n.s.			
Principal Component 1		***	b	a	a	***	a	b	a
Principal Component 2		*	b	a	b	n.s.			

424

425

Table 3. Correlation analysis of soil C and N content (in % of DW), microbial abundances (in % of total PLFAs) and gross rates of protein depolymerization (Protein depol.), microbial amino acid uptake (AA uptake), N mineralization, microbial  $\text{NH}_4^+$  uptake and nitrification, measured with a set of  $^{15}\text{N}$  pool dilution approaches. For correlation with soil C and N content, gross N transformation rates were expressed in  $\mu\text{g N g}^{-1} \text{DW d}^{-1}$ . For correlation with microbial groups, and with each other, gross N transformation rates were corrected for the differences in SOM content between soil horizons and expressed in  $\text{mg N g}^{-1} \text{N d}^{-1}$ . Significance of correlations and correlation coefficients were determined using Spearman's rank correlations. Levels of significance: \*\*\*,  $p < 0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ .

	Protein depol.	AA uptake	N mineralization	$\text{NH}_4^+$ uptake	Nitrification
C	+0.577**	+0.621**	+0.445*	+0.565**	+0.639**
N	+0.519*	+0.541**	+0.418*	+0.539**	+0.650**
Fungi	+0.499*	+0.540**	+0.367	+0.176	+0.428*
Gram negative	+0.368	+0.483*	+0.304	+0.126	+0.528*
Gram positive	-0.305	-0.396	-0.166	-0.092	-0.276
AA uptake	+0.792***				
N mineralization	+0.237	+0.264			
$\text{NH}_4^+$ uptake	+0.356	+0.244	+0.532**		
Nitrification	+0.597**	+0.399	+0.424*	+0.394	

429 **Figure Captions**

430 Fig. 1. Upper panel: Total dissolved N in soil extracts of organic (black bars), cryoturbated (grey bars)  
431 and mineral (white bars) horizons of three tundra sites. Bars are separated into DON (lowest part),  
432  $\text{NH}_4^+$  (middle part) and  $\text{NO}_3^-$  (upper part). Concentrations of  $\text{NH}_4^+$  and DON could not be determined  
433 in the mineral horizons of the tussock tundra site. Lower Panel: Concentrations of inorganic  
434 phosphate in soil extracts of organic (black bars), cryoturbated (grey bars) and mineral (white bars)  
435 horizons of three tundra sites. All bars represent means  $\pm$  standard error. Levels of significance: \*\*\*,  
436  $p < 0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ ; n.s., not significant; n.a., not analyzed (two-way ANOVA or  
437 Kruskal-Wallis test).

438 Fig. 2. Gross rates of protein depolymerization, microbial amino acid uptake, N mineralization,  
439 microbial  $\text{NH}_4^+$  uptake and nitrification (per g total N), of organic (black bars), cryoturbated (grey  
440 bars) and mineral (white bars) horizons of three tundra sites. Rates were measured using a set of  $^{15}\text{N}$   
441 pool dilution approaches. Bars represent means  $\pm$  standard error. Levels of significance: \*\*\*,  $p <$   
442  $0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ ; n.s., not significant; n.a., not analyzed (two-way ANOVA or Kruskal-  
443 Wallis test).

444 Fig. 3. Ordination of organic (black), cryoturbated (grey) and mineral (white) horizons of heath tundra  
445 (Greenland; triangles), tussock tundra (Siberia; circles) and shrub tundra (Siberia; squares) using  
446 Principal Component Analysis. Data include gross rates of protein depolymerization, microbial amino  
447 acid uptake, N mineralization, microbial  $\text{NH}_4^+$  uptake and nitrification (per g total N to correct for  
448 differences in SOM content between horizons), as well as pH values and the relative abundances of  
449 fungi, gram positive and gram negative bacteria (in % of total PLFAs).

450 Fig. 4. Nitrogen use efficiency (NUE) of organic (black bars), cryoturbated (grey bars) and mineral  
451 (white bars) horizons of three tundra sites. NUE was calculated as the proportion of amino acid N  
452 taken up by microorganisms that was not mineralized to  $\text{NH}_4^+$ . Bars represent means  $\pm$  standard

453 error. Levels of significance: \*\*\*,  $p < 0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ ; n.s., not significant (two-way  
454 ANOVA).

455

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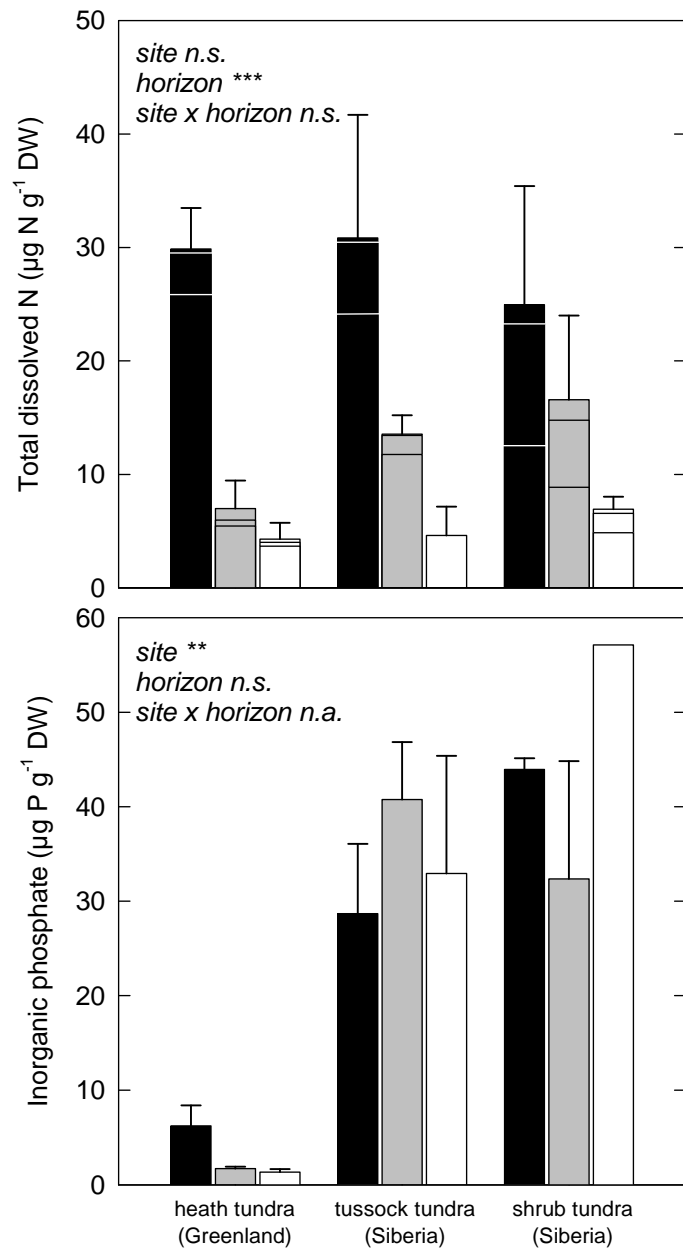


Fig. 1

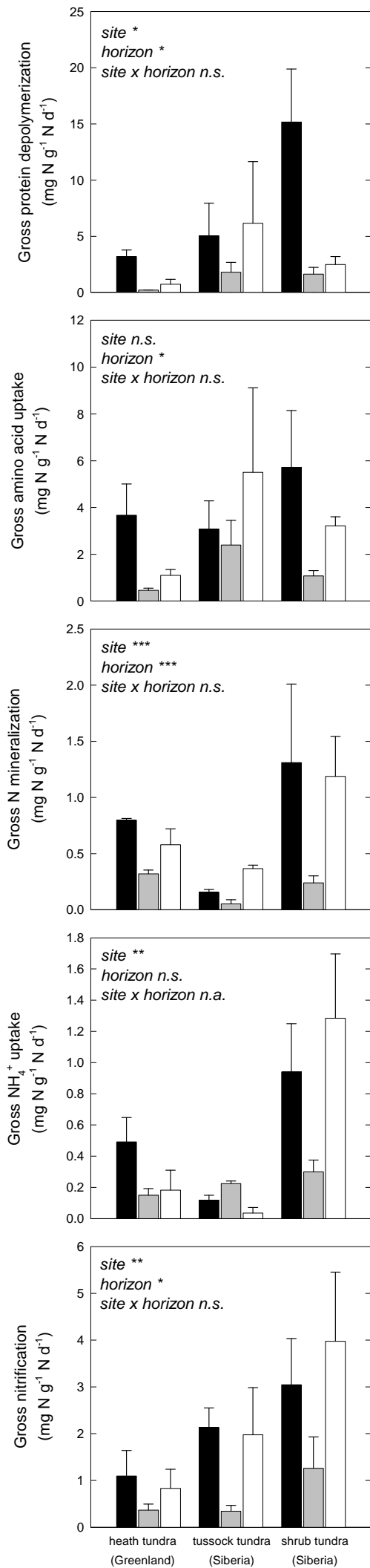


Fig. 2

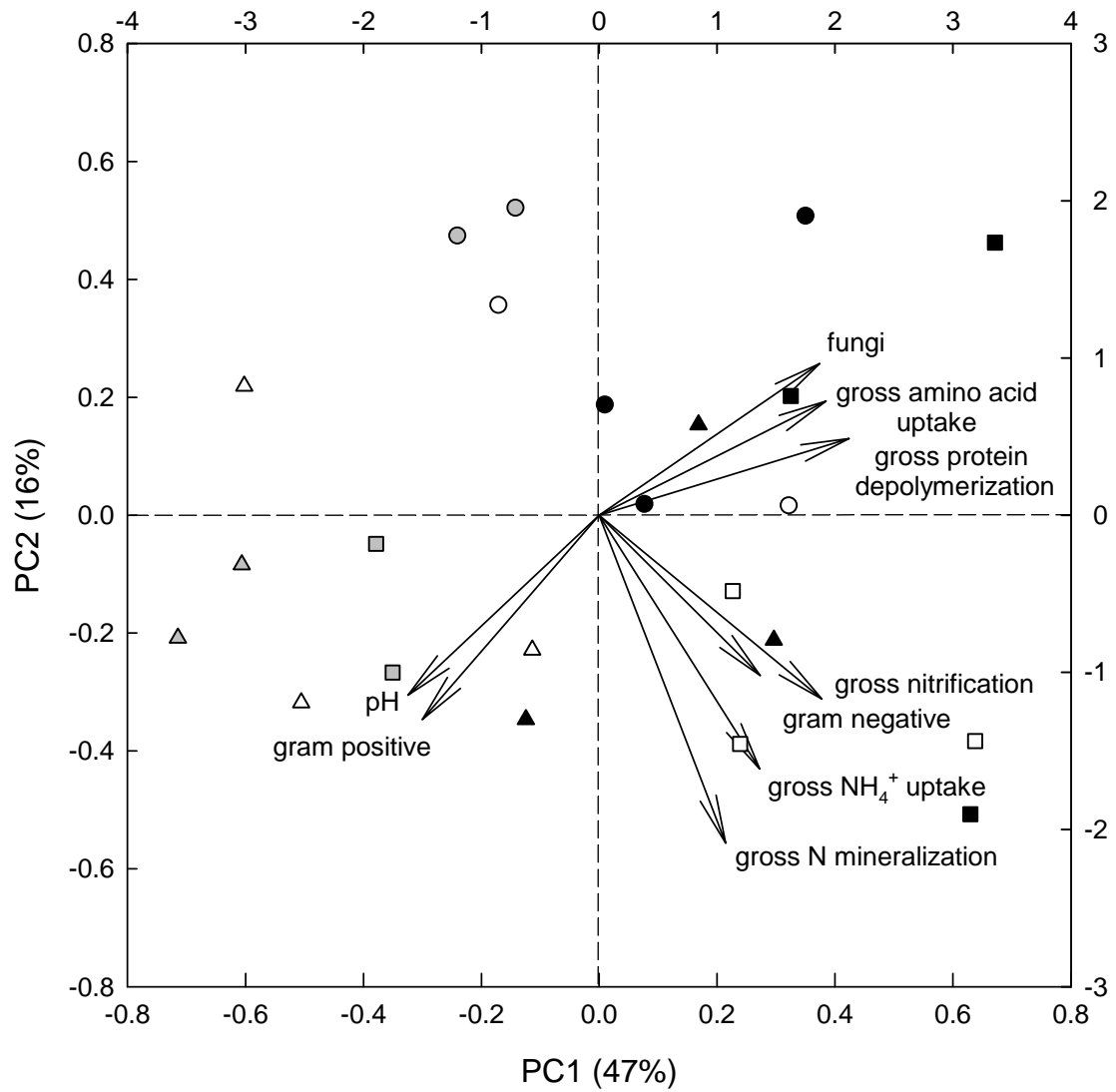


Fig. 3

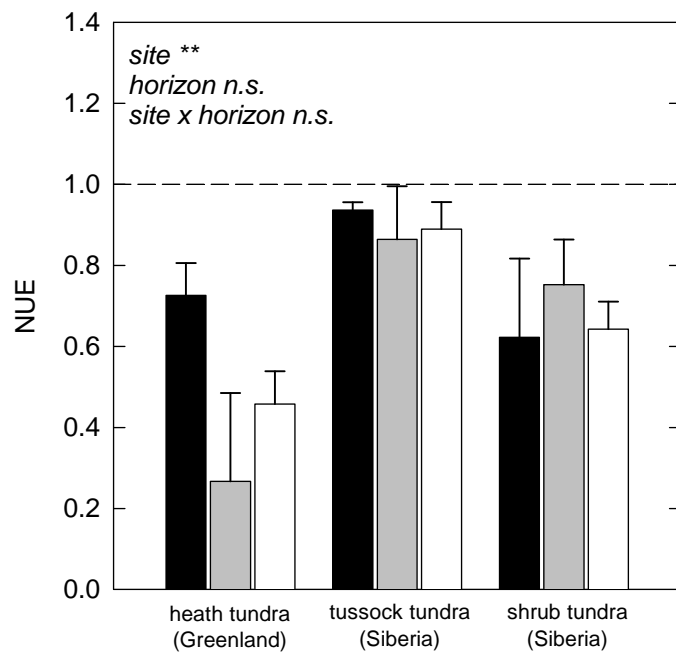
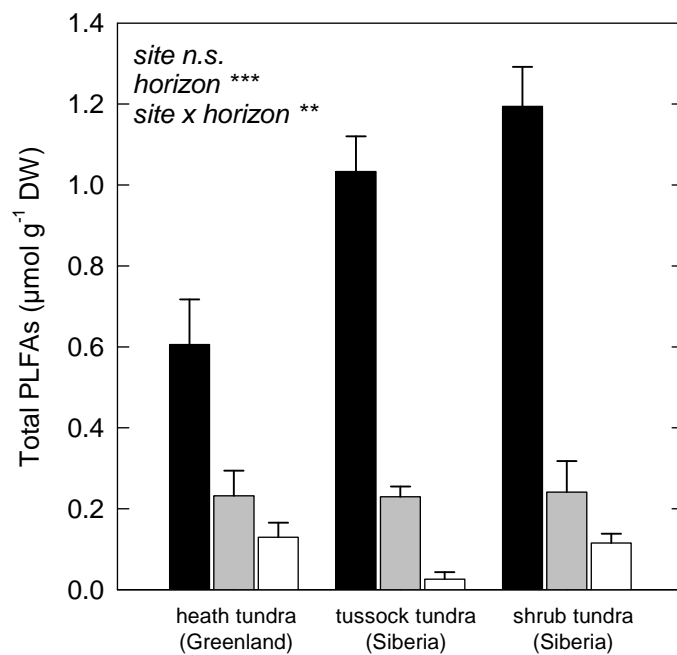
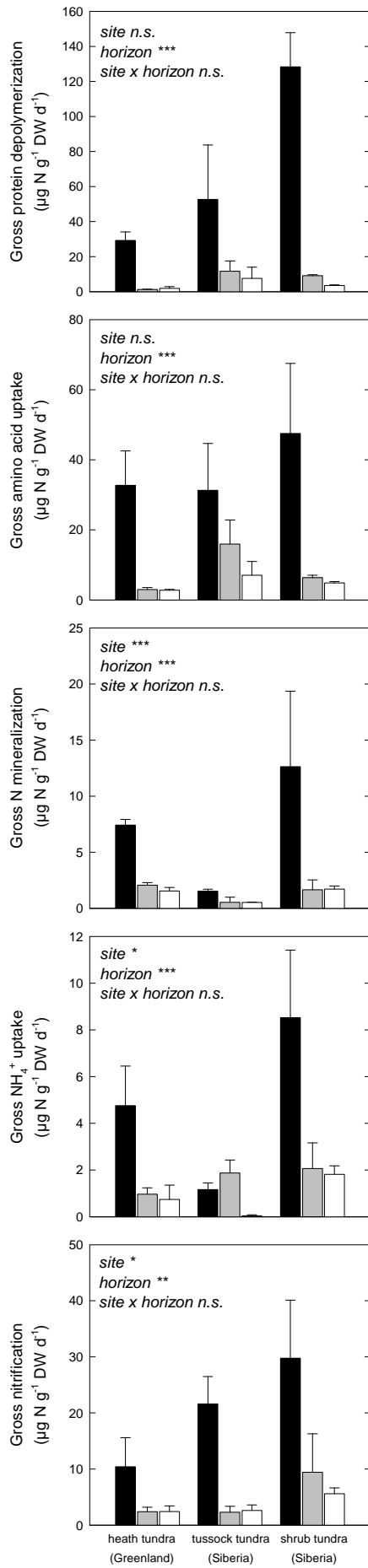


Fig. 4



Supplementary Fig. 1. Concentrations of phospholipid fatty acids (PLFAs) in organic (black bars), cryoturbated (grey bars) and mineral (white bars) horizons of three tundra sites. Bars represent means  $\pm$  standard error. Levels of significance: \*\*\*,  $p < 0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ ; n.s., not significant (two-way ANOVA).



Supplementary Fig. 2. Gross rates of protein depolymerization, microbial amino acid uptake, N mineralization, microbial  $\text{NH}_4^+$  uptake and nitrification (per g dry weight), of organic (black bars), cryoturbated (grey bars) and mineral (white bars) horizons of three tundra sites. Rates were measured using a set of  $^{15}\text{N}$  pool dilution approaches. Bars represent means  $\pm$  standard error. Levels of significance: \*\*\*,  $p < 0.001$ ; \*\*,  $p < 0.01$ ; \*,  $p < 0.05$ ; n.s., not significant (two-way ANOVA).